Phase separated characteristics and soft magnetic properties of [Cu$_{0.6}$FeCrC$_{0.4}$]$_{100-x}$Si$_x$ immiscible composites by laser induction hybrid cladding

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The effect of Si addition on phase separation and soft magnetic properties of [Cu$_{0.6}$FeCrC$_{0.4}$]$_{100-x}$Si$_x$ (x = 3 and 8) immiscible composites produced by laser induction hybrid cladding (LIHC) has been investigated. The duplex structure of immiscible composites was composed of many Fe-rich particles dispersed within Cu-rich upper layer and large amounts of Cu-rich particles embedded within Fe-rich lower layer. However, the increasing Si addition not only induced the precipitation of intermetallic compound Cu$_{6.69}$Si within Cu-rich upper layer, but also increased the area of Fe-rich layer, the size of Fe- and Cu-rich particles, as well as the solubility of Si in Fe- and Cu-rich phases. Moreover, the microhardness of Fe- and Cu-rich layers was increased to 747.3HV$_{0.2}$ and 302.6HV$_{0.2}$ in the Cu$_{55.2}$FeCrC$_{36.8}$Si$_8$ immiscible composite, respectively, which was ~1.6 and ~1.8 times higher than that in the Cu$_{58.2}$FeCrC$_{38.8}$Si$_3$ immiscible composite, due to solid-solution strengthening and dispersion strengthening. Compared to the Cu$_{58.2}$FeCrC$_{38.8}$Si$_3$ immiscible composite, the Cu$_{55.2}$FeCrC$_{36.8}$Si$_8$ immiscible composite presented a saturated magnetization of 13.7 emu/g, relatively lower coercivity of 24.9 Oe and higher Curie temperature of larger than 400 K due to appearance of ferromagnetic a-Fe(Si) phase.

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1. Introduction

When Cu-Fe immiscible alloys are supercooled to a certain temperature below the metastable liquid miscibility gap (T$_{sep}$), the melt can separate into two melts as a result of liquid phase separation: a Fe-rich melt (L1) and a Cu-rich melt (L2) [1,2]. The Cu-Fe immiscible alloys have become potential materials for electronic packaging solders, soft magnetic materials, electrical contact materials, and magnetoresistive materials if the spherical particles as the minor phase can be well dispersed in the metallic matrix [3]. However, the Stokes motion induced by the density difference between the L1 and L2 results in macroscopic segregation due to gravity effect, which limits the wide application of the Cu-Fe immiscible alloys [4].

Generally, once liquid phase separation occurs, the L1 and L2 will not only experience their own supercooling, but also follow a solidification path that differs considerably from that expected for the bulk composition [5]. Much work about the formation mechanism of the LPS has also illustrated that the microstructure of the Cu-Fe immiscible alloys is mainly determined by alloy composition, supercooling, cooling rate and fluid flow during solidification [2,6]. Therefore, many methods have been proposed aimed specifically at improving the microstructure of Cu-Fe immiscible alloys, such as adding different alloy elements (Al [7], Si [8], Ni [9], Nd [10] and Ge [11]) and adopting high supercooling techniques (glass fluxing [12] and electromagnetic levitation [13]). Even under microgravity condition required to overcome the gravity effect on the macroscopic segregation, the collision and coalescence of minority phase...
droplets can not be eliminated completely, implying that the rapid phase growth and coagulation are essentially unavoidable [14].

Compared to conventional solidification techniques, the rapid solidification can obtain cooling rate as high as $10^3$ K/s, which, in turn, can lead to considerable dynamic supercooling, microstructure refinement, and extension of solid solubility [5]. Especially, when Wang et al. [15] produced the core-shell or egg-type microstructure in the Cu-Fe-Si powder by gas atomization, the Cu-Fe immiscible alloys were of particular interest for practical applications because iron had relatively lower cost than other metals such as Ag, Cr and Nb. For example, Koziel et al. [16] and Nagase et al. [17] found that large amounts of spherical fcc-Cu crystalline globules have been precipitated in a melt spin Fe-Cu-based immiscible alloys which contained a Fe-rich amorphous phase. Fu et al. [18] produced the nanostructured Fe$_{50}$Cu$_{40}$ (ratio of atom) immiscible alloy by a combustion synthesis combining rapid solidification, and found that the Fe$_{50}$Cu$_{40}$ immiscible alloy was composed of Fe-rich dendrite (a few micron) embedded in Cu-rich matrix uniformly (~30 nm) with a fractured strength of 1100 MPa and a Vicker’s microhardness of ~2.5 GPa.

Recently, laser induction hybrid cladding (LIHC) was used to produce the Cu-Fe immiscible composites. It was found that the microstructure and properties not only depended on the inoculant addition of CNTs [19], alloying element addition of Al [7], but also on the substrate type and the laser scanning speed [20,21]. Moreover, the previous researches showed that the addition of Si to Cu-Fe immiscible alloys can improve the sensitivity of liquid/liquid phase separation, which results in a wide immiscible gap at temperature as high as 1723 K [8,22]. Therefore, the addition of low Si content (2 wt% [23] and 4 wt% [24]) to Cu-Fe-based immiscible alloys can refine the primary $\alpha$-Fe microstructure and induce the secondary separated phases with high hardness. However, when the addition of Si to Cu-Fe-based immiscible alloys is further increased during LIHC, the microstructure and properties have not been reported. In the present paper, the Si contents (3 wt% and 8 wt%) are added into the Cu-Fe-based immiscible alloys produced by LIHC, the phase separation characteristics and soft magnetic properties are investigated.

2. Experimental procedures

The substrate used was low carbon steel with dimensions of 120 mm $\times$ 60 mm $\times$ 5 mm. The cladding powder was composed of pure copper powder (99.99%), Fe-based alloy powder (Fe-8.5Cr-0.5C) and pure Si powder (99.99%), whose particle size was 40 $\mu$m, 45 $\mu$m and 35 $\mu$m, respectively. The cladding powder with chemical compositions (wt.) of $[\text{Cu}_{0.6}^{\text{Cu}}(\text{FeCrC})_{0.4}]_{100-x}\cdot\text{Si}_x$ ($x = 3$ and 8) were produced by mechanical milling at a rotary speed of 100 rpm for 2 h. The processing apparatus of the laser induction hybrid cladding has been described in detail previously [25]. The processing parameters were as follows: laser powder 5 kW, laser scanning speed 3500 mm/min, spot diameter 4 mm, overlapping rate 50%, powder feeding rate 96.2 g/min, angle between the powder nozzle and the substrate 53°, and the preheated temperature of the substrate 1173 K by adjusting the power of the induction heater. A shield of Ar gas was used to blow the cladding powder into the molten pool and shield the molten pool. The dimensions of the immiscible composites was 90 mm $\times$ 30 mm $\times$ 1.0 mm.

After LIHC, the metallurgical samples were mounted into epoxy resin, polished mechanically, and then etched by a solution with ratio of FeCl$_3$·HCl·H$_2$O = 10 g:25 mL:100 mL. Microstructure was examined by X-ray diffraction (XRD, Brucker D8 Advance) with Cu Kz radiation, scanning electron microscope (SEM, FEI Quanta 200, Netherlands) with energy dispersive X-ray spectroscopy (EDS), and transmission electron microscopy (TEM, JEOL 2100F at 200 kV), respectively. Microhardness of immiscible composite was tested by a Vickers-1000 tester using a load of 1.96 N and a dwelling time of 30 s. Magnetic properties were measured repeatedly by an MPMS XL vibrating sample magnetometer at room temperature with a maximum applied field of 20,000 Oe. The temperature dependence of magnetization curve was determined by heating the immiscible composites in the range of 1.9–400 K at 500 Oe.

3. Results

Fig. 1 shows the results of XRD patterns of the immiscible composites. When the Si addition is 3 wt%, the immiscible composite is mainly composed of face-centered cubic (fcc) $\varepsilon$-Cu and body-centered cubic (bcc) $\alpha$-Fe. When the Si addition is increased to 8 wt%, the immiscible composite consists of fcc $\varepsilon$-Cu, bcc $\alpha$-Fe and Cu$_{50}$Si with a hexagonal structure. Obviously, the increasing Si content results in the intermetallic compound (IMC).

Fig. 2 shows the geometrical profiles of the immiscible composites. Due to liquid phase separation, the duplex structure is formed. According to the results of XRD and EDS, the lower layer is composed of Fe-rich matrix and the upper layer is composed of Cu-rich matrix (Fig. 2a and b). The concave morphologies are also observed at the interface of the lower and upper layers, implying that the violent convection and stirring occur during LIHC. Moreover, when the Si content is increased from 3 wt% to 8 wt%, the area of Fe-rich layer is also increased from 56.6% to 62.7% (Fig. 2c), indicating that the Si content plays an important role in the duplex structure.

Fig. 3 shows the microstructure of the Cu$_{58.2}$(FeCrC)$_{38.8}$Si$_3$.
immiscible composite. According to the results of XRD and EDS (Table 1), the dendrite is identified as α-Fe containing a supersaturated Cu (12.7 wt%) and small amounts of Cr (7.2 wt%) and Si (2.44 wt%). Many spherical Cu-rich particles containing a supersaturated Fe (16.3 wt%) are embedded within the α-Fe interdendrites (Fig. 3a and b) and they have an average diameter of ~3.4 μm (Fig. 4a and b). According to the result of EDS analysis (Table 1), particles in the lower layer. According to the TEM bright and 7.19 wt% Fe, which has a similar composition to the Cu-rich matrix in the upper layer (Fig. 4c). The Cu-rich matrix in the upper layer contains 92.81 wt% Cu (7.11 wt%), 6.75 wt% Cr and large amounts of Si (10.36 wt%). These Cu-rich particles have a relatively larger diameter (~6.8 μm) compared to that in the Cu58.2(FeCrC)38.8Si3 immiscible composite (Fig. 5). However, large amounts of spherical Fe-rich particles are dispersed within the Cu-rich matrix in the upper layer (Fig. 3c). These Fe-rich particles containing a supersaturated Cu (16.34 wt%), small amounts of Cr (5.77 wt%) and Si (3.25 wt%) have an average diameter of ~3.5 μm (Fig. 5). Additionally, large amounts of white nano-scaled Cu-rich grains are precipitated inside the Fe-rich particles due to the secondary liquid phase separation [5], as shown in Fig. 3d. The Cu-rich matrix in the upper layer contains 92.81 wt% Cu and 7.19 wt% Fe, which has a similar composition to the Cu-rich particles in the lower layer. According to the TEM bright field image and selected area electron diffraction (SAED) patterns (Fig. 3e and f), the nano-scaled Fe-rich particle is confirmed to be α-Fe phase with (220) and (024) planes and the Cu-rich matrix is confirmed to be ε-Cu phase with (002) and (020) planes.

Fig. 4 shows the microstructure of the Cu55.2(FeCrC)36.8Si8 immiscible composite. Large amounts of spherical Cu-rich particles are also dispersed within the α-Fe interdendrites in the lower layer (Fig. 4a and b). According to the result of EDS analysis (Table 1), these Cu-rich particles contain 11.47 wt% Fe, 2.27 wt% Cr and 5.04 wt% Si. The α-Fe dendrites also contain a supersaturated Cu (7.11 wt%), 6.75 wt% Cr and large amounts of Si (10.36 wt%). These Cu-rich particles have a relatively larger diameter (~6.8 μm) compared to that in the Cu58.2(FeCrC)38.8Si3 immiscible composite (Fig. 5). However, large amounts of impinging Fe-rich particles are dispersed within the Cu-rich matrix in the upper layer (Fig. 4c). The Fe-rich particle contains a supersaturated Cu (7.81 wt%), small amounts of Cr (9.52 wt%) and large amounts of Si (17.44 wt%) and has a diameter of ~10 μm. The Cu-rich matrix contains a supersaturated Fe (4.26 wt%) and small amounts of Si (2.89 wt%). Interestingly, large amounts of white petal-like precipitations are formed on the Cu-rich matrix (Fig. 4d). These precipitations contain 65.41 wt% Cu, 24.21 wt% Fe, 3.21 wt% Cr and 7.08 Si. According to the TEM bright field image and SAED patterns (Fig. 4e and f), the petal-like precipitation is confirmed to be hexagonal Cu62Si with (010) and (111) planes, the spherical Fe-rich particle is bcc α-Fe with (002) and (111) planes and the Cu-rich matrix is fcc ε-Cu phase with (111) and (111) planes.

Fig. 6 shows the microhardness distribution of the immiscible composites and pure Cu produced by LIHC. The immiscible composites have a sharply gradient distribution of microhardness, which is consistent with the duplex structure. For the Cu58.2(FeCrC)38.8Si3 immiscible composite, the Fe-rich lower layer has an average microhardness of 472.8HV0.2, which is ~2.8 times higher than the Cu-rich upper layer (165.5HV0.2). The microhardness of the Cu-rich layer is higher than that of pure Cu produced using LIHC (104HV0.2). When the Si content is increased to 8 wt%, the microhardness of the Fe- and Cu-rich layers is increased to 747.3HV0.2 and 302.6HV0.2, respectively, which is ~1.6 and ~1.8 times higher than that of Fe- and Cu-rich layers in the Cu58.2(FeCrC)38.8Si3 immiscible composite, respectively. Obviously, the increasing Si content can enhance the microhardness of the immiscible composites significantly.

Fig. 7 shows the magnetic hysteresis loops of the immiscible composites at 300 K. For the Cu58.2(FeCrC)38.8Si3 immiscible composite, the magnetization can not reach a saturated state when the applied field is up to 20,000 Oe. The maximum magnetization is only ~2.3 emu/g and the coercivity is ~166.9 Oe (Fig. 7a). For the Cu55.2(FeCrC)36.8Si8 immiscible composite, the saturated magnetization (~13.7 emu/g) can be obtained sharply when the applied field is only larger than 5000 Oe. The coercivity is only ~24.9 Oe (Fig. 7b). It indicates that the increasing Si content can also improve the soft magnetic properties of the immiscible composites.
4. Discussion

It has shown that the cooling rate of LIHC is of order of $10^4\sim10^5$ K/s [21,25], which, in turn, can lead to extension of solid solubility, such as a supersaturated Cu in Fe or a supersaturated Fe in Cu, and considerable dynamic supercooling of ~150 K. Generally, the supercooling of the Cu-Fe immiscible alloys containing 40~65 wt% Cu is only 80~100 K if the liquid phase separation occurs [26]. Therefore, the Cu58.2(FeCrC)38.8Si3 and Cu55.2(FeCrC)36.8Si8 immiscible alloys can be supercooled readily below the $T_{sep}$ and separate into the two melts: a Fe-rich melt as the minor phase (L1) and a Cu-rich melt as the major phase (L2). Subsequently, the L1 and L2 will follow their own solidification path to finish nucleation and growth. Generally, the critical nucleation work $\Delta G^*$, can be given as follows [27]:

<table>
<thead>
<tr>
<th>Immiscible composites</th>
<th>Cu-rich particle</th>
<th>Fe-rich matrix</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cu58.2(FeCrC)38.8Si3</td>
<td>Fe  Cr  Si Cu  Cr  Si Fe</td>
<td>16.3  0  0 Bal. 12.7  7.2  2.44 Bal.</td>
</tr>
<tr>
<td>Cu-rich particle</td>
<td>Fe  Cr  Si Cu  Cr  Si Cu</td>
<td>16.34  5.77  3.25 Bal. 7.19  0  0 Bal.</td>
</tr>
<tr>
<td>Fe-rich particle</td>
<td>Cu  Cr  Si Fe  Cr  Si Cu</td>
<td>7.81  9.52  17.44 Bal. 4.26  0  2.89 Bal.</td>
</tr>
<tr>
<td>Cu55.2(FeCrC)36.8Si8</td>
<td>Fe  Cr  Si Cu  Cr  Si Fe</td>
<td>11.47  2.27  5.04 Bal. 7.11  6.75  10.36 Bal.</td>
</tr>
<tr>
<td>Cu-rich particle</td>
<td>Cu  Cr  Si Fe  Cr  Si Cu</td>
<td>7.81  9.52  17.44 Bal. 4.26  0  2.89 Bal.</td>
</tr>
<tr>
<td>Fe-rich particle</td>
<td>Cu  Cr  Si Fe  Cr  Si Cu</td>
<td>7.81  9.52  17.44 Bal. 4.26  0  2.89 Bal.</td>
</tr>
</tbody>
</table>
\[ \Delta G' = \frac{16\pi}{3} \frac{\sigma^3}{\Delta G_v} f(\theta) \] (1)

where \( \sigma \), \( \Delta G_v \) and \( f(\theta) \) are the interfacial energy, the Gibbs free energy difference between solid and liquid, and catalytic factor for heterogeneous nucleation, respectively. According to the Cu-Fe binary phase diagram [28], the L1 has a relatively lower formation work for nucleation \( \Delta G' \) and has a good wettability with the Fe-substrate [20], the L1 can sink towards the bottom of the melt pool and solidify before the L2. As a result, the Fe-rich layer will form at the bottom of the melt pool and the Cu-rich layer will solidify on the surface of the solidified Fe-rich layer, resulting in the duplex structure (Fig. 2a and b).

During LIHC, the laser energy is consumed in three aspects: (1) melting the composite powder; (2) melting the substrate to form the melt pool; and (3) thermal radiation and the other loss. An energy balance for LIHC can expressed as follows [29,30]:

\[ P = P_p + P_s + P_L \] (2)

where \( P \) is the laser power, \( P_p \) is the power used to melt the composite powder, \( P_s \) is the power used to melt the substrate to form the melt pool, and \( P_L \) is the power lost by reflection, radiation, convection, and etc. Generally, the increasing Si content can decrease the melting point of composite powder, resulting to a decrease in \( P_p \). Correspondingly, the \( P_s \) is increased, which in turn leads to an increase in melting height of substrate. As a result, the
dilution of the immiscible composite is increased from 8.6% to 13.5% so that the more melting of the substrate increases the area of Fe-rich layer (Fig. 2c).

Moreover, once the LPS occurs during LIHC, the L2 as the minor phase can shrink into large amounts of liquid droplets in the Fe-rich lower melt due to minimization surface energy. Similarly, the L1 as the minor phase can also shrink into many liquid droplets in the Cu-rich upper melt. These liquid droplets can move in the liquid matrix, driven by Marangoni motion and Stokes motion. Previous results [15] have shown that the velocity of the liquid droplets induced by Marangoni motion is $-10^4$ times faster than that induced by Stokes motion, implying that the Stokes effect can be ignored. For a simple analysis, the velocity of the liquid droplets induced by Marangoni motion is estimated as follows [15]:

$$v_m = \frac{2r_d}{3(3\eta_d + 2\eta_m)} \frac{\partial \sigma}{\partial T} G$$

(3)

where $r_d$ is the radius of the liquid droplet, $G$ is the temperature gradient, $\eta_d$ and $\eta_m$ are the viscosities of the liquid droplet and the liquid matrix, respectively, and the $\frac{\partial \sigma}{\partial T}$ is the interfacial energy. Furthermore, the viscosity of melt can be described as follows [31]:

$$\eta = 5.7 \times 10^{-6} \sqrt{\frac{M \cdot T_m}{V_m}}$$

(4)

where $M$ is the atom weight, and $V_m$ is the atom volume at temperature $T_m$. According to the Eq. (4), the viscosity of Si calculated (2.24 cP) is smaller than that of Fe (4.55 cP) and Cu (3.95 cP). Thus, the increasing Si content can decrease the viscosity of melt during LIHC. The velocity of liquid droplets driven by Marangoni motion is increased according to the Eq. (3). Thus, the increasing Si content can obtain higher Marangoni velocity of liquid droplets in the liquid matrix. Li et al. [32] also reported a similar effect of the Ce addition on Marangoni velocity of the Cu-Sn droplets in the Cu-Sn-Bi immiscible alloy. This in turn increases the probability of collision and coalescence of liquid droplets in the liquid matrix. Therefore, the liquid droplets can collide and impinge to form a new larger droplet by the diffusion-coupling mechanism or the gradient induced coupling mechanism [33]. As a result, the Fe- and Cu-rich particles in the Cu$_{55.2}$FeCr$_{36.8}$Si$_{8}$ immiscible composite present a relatively larger size compared to those in the Cu$_{58.2}$FeCr$_{38.8}$Si$_{3}$ immiscible composite (Figs. 3–5).

Furthermore, the ε-Cu and α-Fe phases present the index of crystallographic direction [100] and [221] in the Cu$_{55.2}$FeCr$_{36.8}$Si$_{8}$ immiscible composite, respectively. However, they present the index of crystallographic direction [110] and [310] in the Cu$_{58.2}$FeCr$_{38.8}$Si$_{3}$ immiscible composite, respectively. Generally, the solid solubility of the Cu in α-Fe phase is negligible at room
temperature [26]. Due to high cooling rate during LIHC, the solute trapping can occur [21], resulting in a supersaturated Cu in the $\alpha$-Fe phase and a supersaturated Fe in the $\epsilon$-Cu phase. However, as a result of increasing Si content, the supersaturated solid solubility of Cu in the $\alpha$-Fe particles and matrix decreases from 16.34 wt% and 12.7 wt% to 7.81 wt% and 7.11 wt%, respectively, which is in reasonable accordance with the experimental results from Zhao et al. [8]. Similarly, the supersaturated solid solubility of Fe in the $\epsilon$-Cu particles and matrix also decreases from 16.3 wt% to 11.47 wt% and 4.26 wt%, respectively. However, the solid solubility of Si in the $\alpha$-Fe particles and matrix decreases significantly with the increasing of Si addition. This in turn causes the lattice distortion and non-coherent interface of $\alpha$-Fe and $\epsilon$-Cu phases, resulting in different index of crystalline direction and solid solution strengthening. Additionally, large amounts of $\alpha$-Fe and $\epsilon$-Cu particles are dispersed in the $\epsilon$-Cu and $\alpha$-Fe matrix, resulting in dispersion strengthening. Especially, due to the precipitation of intermetallic compound (IMC) Cu$_{6.69}$Si precipitations and an increase in diameter of the $\epsilon$-Cu and $\alpha$-Fe particles, the microhardness of the Cu$_{58.2}$Fe$_{38.8}$Si$_3$ immiscible composite is strengthened significantly compared to the Cu$_{58.2}$Fe$_{38.8}$Si$_3$ immiscible composite and pure Cu produced by LIHC (Fig. 6).

When the Si content is only 3 wt%, the immiscible composite cannot present a saturated magnetization (Fig. 7a). This is because the presence of the paramagnetic contribution of the $\epsilon$-Cu phase, which has a similar magnetic characteristic to Fe$_{50}$Si$_{50}$ nanostructured powder produced by a combination of melting and milling process [34]. Especially, when the immiscible composite is heated in the range of 1.9–400 K at 500 Oe, the magnetization of the immiscible composite is reduced significantly to zero at ~200 K shown in Fig. 8a, conforming that the ferromagnetic (FM)-paramagnetic (PM) transformation at Curie temperature lower than 200 K. However, when the Si content is increased to 8 wt%, the immiscible composite presents a characteristic of the saturated magnetization, many be because increasing Si content can increase the area of the FM $\alpha$-Fe(Si) layer containing 10.36 wt% Si and the diameter of the FM $\alpha$-Fe(Si) particles containing 17.44 wt% Si, which can also be confirmed by Fig. 8b as a result of almost unchanged magnetization during heating. This is because substitution of Cu atoms by Si atoms in the bcc Fe-rich matrix or particles and the magneto-volume effect results in positive exchange interactions between Fe atoms [35]. The decrease in coercivity in the Cu$_{55.2}$Fe$_{36.8}$Si$_8$ immiscible composite are due to absent precipitation of nano-sized Fe-rich particles in the Cu-rich matrix [36]. Obviously, the increasing Si addition can improve the Curie temperature (higher than 400 K) of the immiscible composites.

5. Conclusions

(1) The duplex structure of the Cu$_{58.2}$Fe$_{38.8}$Si$_3$ immiscible composite is mainly composed of many $\epsilon$-Cu particles embedded within the $\alpha$-Fe lower dendrites and large amounts of $\alpha$-Fe particles dispersed within the $\epsilon$-Cu upper matrix. When the Si addition is increased to 8 wt%, the immiscible composite has a similar duplex structure, while large amounts of intermetallic compounds identified as Cu$_{6.69}$Si are precipitated within the $\epsilon$-Cu upper layer.

(2) The increasing Si addition to Cu-Fe-based immiscible composites can increase the area of Fe-rich layer, the size of Fe- and Cu-rich particles, the solid solubility of Si in the $\alpha$-Fe and $\epsilon$-Cu phases, while decrease the supersaturated solid solubility of Cu and Fe in the $\alpha$-Fe and $\epsilon$-Cu phases. As a result, the microhardness of the Fe- and Cu-rich layers is improved significantly, due to the IMC Cu$_{6.69}$Si precipitations, and an increase in diameter of the $\epsilon$-Cu and $\alpha$-Fe particles.

(3) The Cu$_{58.2}$Fe$_{38.8}$Si$_3$ immiscible composite cannot present a saturated magnetization and is characterized by ferromagnetic (FM)-paramagnetic (PM) transformation during heating. However, the Cu$_{55.2}$Fe$_{36.8}$Si$_8$ immiscible composite presents the saturated magnetization (13.7 emu/g), relatively lower coercivity (24.9 Oe) and higher Curie temperature (larger than 400 K) as a result of the ferromagnetic $\alpha$-Fe(Si) phase.

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