



# Effect of $C_2H_2/N_2$ partial pressure ratio on microstructure and mechanical properties of Ti-Al-Si-C-N coatings

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## ABSTRACT

Ti-Al-Si-C-N nanocomposite coatings were deposited on cemented carbide (WC-10 wt%, Co) substrates by reactive magnetron sputtering technique. The effect of  $C_2H_2/N_2$  partial pressure ratio on element concentration, deposition rate, microstructure, cross-sectional morphology, hardness and tribological properties of the coatings were studied. The coatings were found to have a nanocomposite structure consisting of nanocrystallinities (Ti,Al) (C,N) and amorphous phase. The coating's columnar crystals structure was restrained with the increase of  $C_2H_2/N_2$  partial pressure ratio. The deposition rates and nanohardness of coatings firstly increased with the increase of  $C_2H_2/N_2$  partial pressure ratio and reached their maximum values of 133 nm/min and 35 Gpa, respectively, at  $C_2H_2/N_2$  partial pressure ratio of 1:2 before decreasing. The friction coefficient and wear rates of Ti-Al-Si-C-N coatings dramatically decreased with the increase of  $C_2H_2/N_2$  partial pressure ratio and reached the lowest value of 0.17 and  $1.1 \times 10^{-6} \text{ mm}^3/\text{Nm}$ , respectively, at  $C_2H_2/N_2$  partial pressure ratio of 1:1.

## 1. Introduction

Due to the excellent properties such as high hardness, good wear resistance and chemical stability, TiN coatings as one of the protective hard coatings in cutting tools, dies and many mechanical components to improve their lifetime and performance [1–5]. However, one of the major the main drawbacks of TiN is its limited oxidation resistance (approximately 500 °C) [6]. To overcome this problem, aluminum was alloyed to form a TiAlN nanocrystalline coating, which significantly improved the high temperature hardness and oxidation resistance up to 800 °C [7–10]. It has also been reported that incorporation of silicon in TiN coating can further improve hardness (> 40 GPa), oxidation resistance (> 800 °C) and wear resistance of the TiSiN coating due to the microstructure of nano-sized TiN crystals embedded in amorphous  $Si_3N_4$  [11–15]. Therefore, the possibility of combining hard phases such as nanocrystalline TiAlN with improved oxidation resistance, with  $Si_3N_4$  amorphous phase achieving superhardness, has allowed to constitute the TiAlSiN coating, which significantly increases the life time and performance of high-speed cutting tools [16–20].

Moreover, incorporating the carbon element has been found to significantly improve the tribological property of TiN-based coatings [21–28], which usually lack solid lubricating properties and exhibit

relatively high friction coefficients (0.4–0.9) [29–32]. Hydrocarbon gases such as  $CH_4$  and  $C_2H_2$  are usually used as carbon source in preparing those coatings. Therefore, the partial pressure ratio between carbon-containing gas and  $N_2$  becomes an important control parameter that affects the properties of resulting coatings. For example, Huang et al. [33] deposited  $TiC_xN_{(1-x)}$  coatings by filtered arc and found that a significant increase in surface roughness, microhardness and wear resistance of the coatings with the increase of  $CH_4/N_2$  gas flow rate ratio. Jang et al. [34] indicated that tribological property of  $TiAlC_xN_{1-x}$  coatings remarkably improved with the increase of  $CH_4/(CH_4 + N_2)$  gas flow rate ratio. Xie et al. [35] synthesized Ti-Al-Si-C-N coatings by multi-plasma immersion ion implantation with different  $C_2H_2/N_2$  gas flow rate ratios and pointed out that a-C is found in the coating and acts as a solid lubricant decreasing friction and wear. Those results from previous studies indicated that the microstructure and properties of the TiN-based hard coatings are significantly dependent on the hydrocarbon/nitrogen partial pressure ratio. However, the effect of  $C_2H_2/N_2$  partial pressure ratio on the microstructure and mechanical properties of the Ti-Al-Si-C-N coatings synthesized by reactive magnetron sputtering technique is still not clear.

In this study, Ti-Al-Si-C-N hard coatings were deposited by reactive magnetron sputtering technique with different  $C_2H_2/N_2$  partial pressure

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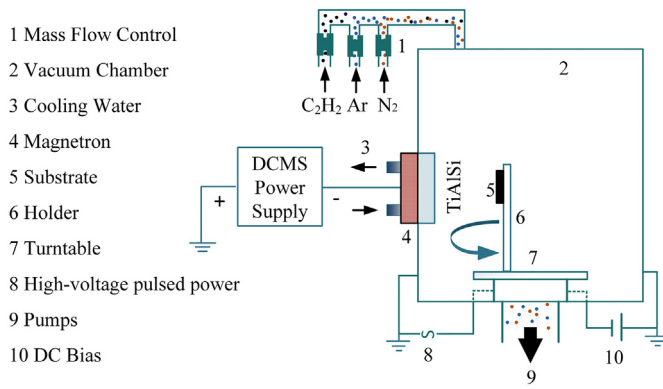


Fig. 1. Schematic of the deposition apparatus.

ratios. The microstructure, morphology, nanoindentation hardness, and tribological properties of these coatings were studied as a function of  $C_2H_2/N_2$  partial pressure ratio. The mechanism of how the  $C_2H_2/N_2$  partial pressure ratio affects the coatings' microstructure and mechanical properties was discussed.

## 2. Experimental

### 2.1. Sample preparation

Ti-Al-Si-C-N coatings were prepared by reactive magnetron sputtering technique with different  $C_2H_2/N_2$  partial pressure ratios. As shown in the schematic in Fig. 1, the rectangular ( $12 \times 10 \text{ cm}^2$ )  $Ti_{0.64}Al_{0.3}Si_{0.06}$  targets were installed on the vacuum chamber wall. A detailed description of the system can be found in our earlier works [19, 20]. The deposition of Ti-Al-Si-C-N coatings was carried out in a mixture of high purity argon (99.999%), acetylene ( $C_2H_2$  99.999%) and nitrogen ( $N_2$  99.999%). The total working pressure and partial pressure of argon were kept constant at 0.8 Pa and 0.6 Pa, respectively. To study the effects of  $C_2H_2/N_2$  partial pressure ratio, several independent mass flow controls were used to adjust  $C_2H_2/N_2$  partial pressure ratio ranging from 0:1, 1:4, 1:3, 1:2 to 1:1. The cemented carbide (WC-10 wt% Co) was used as the substrate. Prior to deposition, all substrates were firstly mechanically polished, then ultrasonically cleaned, and etched by Ar ions directly to remove surface contaminants and the surface oxidation layer. The temperature of the substrates was kept constant at  $150^\circ\text{C}$  during deposition. To enhance the adhesion of the deposited coating to the substrate, high-energy ion bombardment with average nitrogen ion energy of about 10 keV was applied for 20 min prior to the deposition. The sputtering parameters for Ti-Al-Si-C-N coatings are listed in Table 1. During deposition the substrate current stayed around  $0.15 \pm 0.01 \text{ A}$ .

### 2.2. Characterization and evaluation of the coatings

The element content, cross-sectional morphology and thickness of coatings were measured by energy dispersive spectroscopy (EDS) and scanning electron microscope (SEM: Zeiss Supra 55). The crystalline structure of the coatings was characterized by X-ray diffraction (XRD:

D/Max 2500) with  $CuK\alpha$  radiation in grazing-incidence mode ( $2^\circ$ ). Raman spectra were studied using a HR-800 Raman spectrometer operated at a laser wavelength of 532 nm to characterize the substructures of free carbon in the coatings. The hardness and Young's modulus of the coatings were obtained by nanoindentation (Nano-Indenter G200, Agilent) using continuous stiffness method (CSM). A maximum load of 50 mN was used to assure that the indentation depth was within the 5–10% of the coating thickness.

The tribological performance of coatings was evaluated by a HT-600 ball-on-disk tribometer using a load of 3.4 N. GCr15 bearing steel balls (6 mm diameter) were used as the counterparts. The tests were carried out at a sliding speed of 0.1 m/s under dry conditions at room temperature without lubricants. The sliding radius was 2 mm and the sliding time was 30 min. After the tests, wear tracks were investigated by a laser confocal scanning microscope (OLYMPUS OLS4100, Japan). Depth profiles across the width of the track, perpendicular to the sliding direction, were also calculated.

## 3. Results and discussion

### 3.1. Microstructure of the Ti-Al-Si-C-N coating

Table 2 shows the composition of the Ti-Al-Si-C-N coatings deposited with different  $C_2H_2/N_2$  partial pressure ratios. Not surprisingly, as  $C_2H_2/N_2$  partial pressure ratio increased, the C contents increase from 0 at.% to 59.48 at.% while N contents decrease from 37.95 at.% to 12.48 at.%. Meanwhile, the Ti, Al and Si contents also decrease correspondingly. However, the Ti, Al and Si relative content ratio in the coating is approximately 5.7:4.3:1, which is remains basically unchanged.

The cross-sectional morphologies of Ti-Al-Si-C-N coatings with different  $C_2H_2/N_2$  partial pressure ratios were observed by SEM and the results were illustrated in Fig. 2. The thickness of the Ti-Al-Si-C-N coatings with the  $C_2H_2/N_2$  partial pressure ratio of 0:1, 1:4, 1:3, 1:2, and 1:1 are 2.65, 2.95, 3.4, 4.0, and 3.0  $\mu\text{m}$ , respectively. As shown in Fig. 2a, the columnar crystal structure is clearly seen in coatings deposited at  $C_2H_2/N_2$  partial pressure ratio of 0:1. As  $C_2H_2/N_2$  partial pressure ratio increased to 1:4, the columnar crystal structure is still visible, as shown in Fig. 2b. However, the length and the diameter of the columns both decrease comparing to the TiAlSiN coating shown in Fig. 2b. In the coating deposited with the  $C_2H_2/N_2$  partial pressure ratio of 1:3, columnar crystal structure disappears and the coating shows a granular structure instead, as shown in Fig. 2c. As  $C_2H_2/N_2$  partial pressure ratio further increased to 1:2, neither columnar crystal structure nor granular structure can be seen, and the coating exhibits a densified microstructure with a smooth cross-section, as shown in Fig. 2d. This densified microstructure is consistent with the nanocomposite structure found in previous studies [36]. Finally, when  $C_2H_2/N_2$  partial pressure ratio increased to 1:1, the cross-section of the deposited coating shows an amorphous like morphology, as shown in Fig. 2e. It is indicated that with the increase of  $C_2H_2/N_2$  partial pressure ratio, the columnar crystal growth is restrained. This is likely due to the incorporation of the amorphous carbon or amorphous  $Si_3N_4$ , located at the grain boundary, can hinder the migration of grain boundary [28].

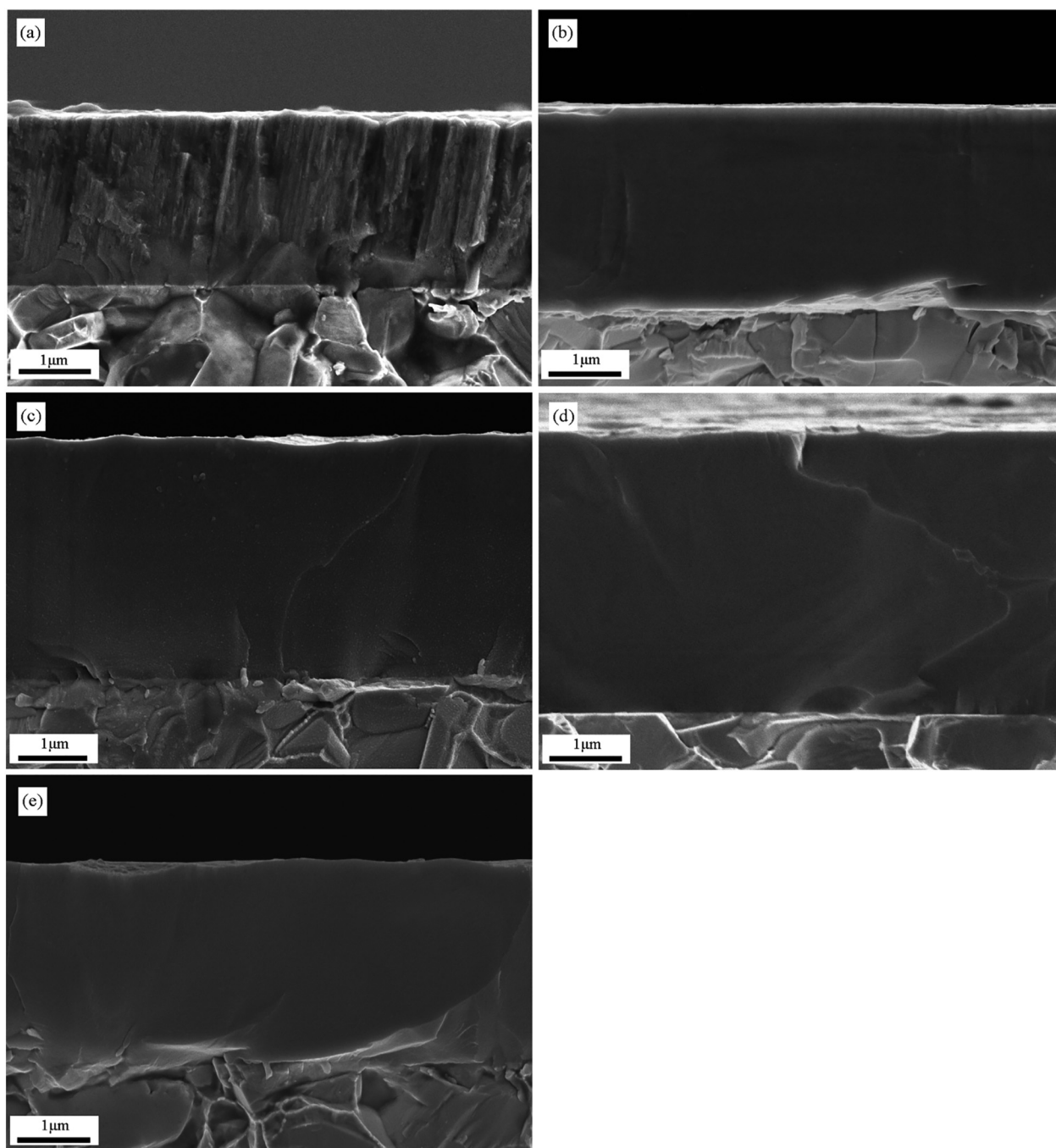
Fig. 3 presents the deposition rates of Ti-Al-Si-C-N coatings as a function of  $C_2H_2/N_2$  partial pressure ratio. When  $C_2H_2/N_2$  partial

Table 1  
Details of deposition parameters.

Sample	Target-substrate distance (cm)	Substrate bias (V)	Temperature ( $^\circ\text{C}$ )	$P_{C_2H_2}/P_{N_2}$	$P_{Ar}$ (Pa)	Working pressure (Pa)	Target current (A)
0: 1	8	−100	150	0/1	0.6	0.8	2
1: 4	8	−100	150	1/4	0.6	0.8	2
1: 3	8	−100	150	1/3	0.6	0.8	2
1: 2	8	−100	150	1/2	0.6	0.8	2
1: 1	8	−100	150	1/1	0.6	0.8	2

**Table 2**  
Composition of Ti-Al-Si-C-N coatings.

Sample	Element composition (at.%)				
	Ti	Al	Si	C	N
0:1	32.03 $\pm$ 0.20	24.43 $\pm$ 0.15	5.59 $\pm$ 0.06	0	37.95 $\pm$ 0.39
1:4	29.28 $\pm$ 0.31	22.25 $\pm$ 0.28	5.17 $\pm$ 0.10	14.57 $\pm$ 0.12	28.73 $\pm$ 0.56
1:3	28.98 $\pm$ 0.45	19.94 $\pm$ 0.26	4.96 $\pm$ 0.19	18.20 $\pm$ 0.25	27.92 $\pm$ 0.95
1:2	25.94 $\pm$ 0.50	18.99 $\pm$ 0.31	4.52 $\pm$ 0.09	23.61 $\pm$ 0.25	26.94 $\pm$ 0.42
1:1	14.13 $\pm$ 0.19	11.49 $\pm$ 0.22	2.42 $\pm$ 0.03	59.48 $\pm$ 0.28	12.48 $\pm$ 0.43



**Fig. 2.** Cross-sectional morphologies of Ti-Al-Si-C-N coatings with different  $C_2H_2/N_2$  partial pressure ratios: (a) 0:1 (b) 1:4 (c) 1:3 (d) 1:2 (e) 1:1.

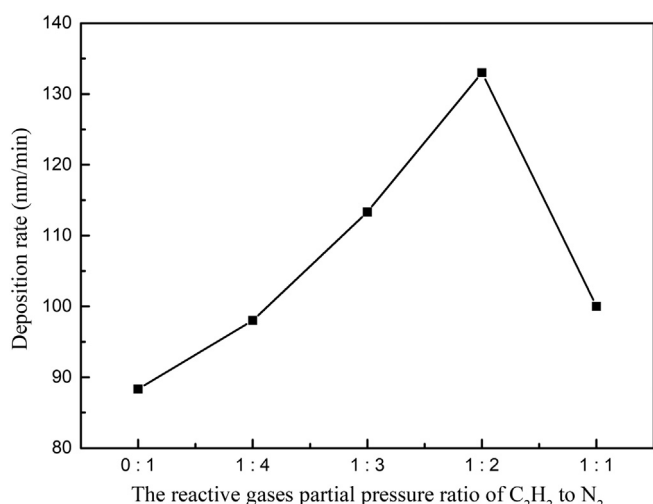


Fig. 3. Deposition rates of Ti-Al-Si-C-N coatings as a function of the C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio.

pressure ratio increased from 0:1 to 1:2, the deposition rate dramatically increased from 88.33 nm/min to about 133 nm/min. This phenomenon can be explained as follows: The ionization energy for N<sub>2</sub> and C<sub>2</sub>H<sub>2</sub> are 15.58 eV [37] and 11.403 eV [38, 39], respectively. Therefore, C<sub>2</sub>H<sub>2</sub> are more apt to be ionized than N<sub>2</sub>. In other word, C element is more active during the deposition. Carbon-containing particles are more likely to settle down and form carbon-based coatings on the substrate. As a result, the deposition rates increased with the increase of C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio. With the further increasing C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio, the deposition rate decreased. This is due to the fact that TiAlSi targets were partial covered with amorphous carbon or even carbide, which in turn reduced the amount of atoms and ions sputtered from the targets, leading to the occurrence of poisoning phenomenon [7]. Those results indicated that the deposition rate of Ti-Al-Si-C-N coatings depended not only on the discharge parameters but also on the C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio. Similar results also were reported in Refs [40].

Fig. 4 shows the XRD patterns of the Ti-Al-Si-C-N coatings with different C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratios, and the standard peaks of TiN, TiC, and substrate WC are tagged. The diffraction peaks of Si<sub>3</sub>N<sub>4</sub>, SiC, TiSi crystalline phases were not discovered, indicating that the silicon atoms likely exist in the form of amorphous silicon nitride (a-Si<sub>3</sub>N<sub>4</sub>) and

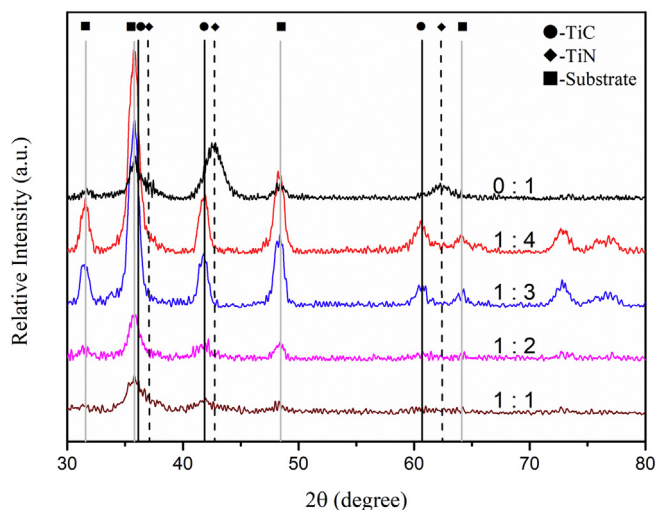


Fig. 4. X-ray diffraction pattern of Ti-Al-Si-C-N coatings with different C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratios.

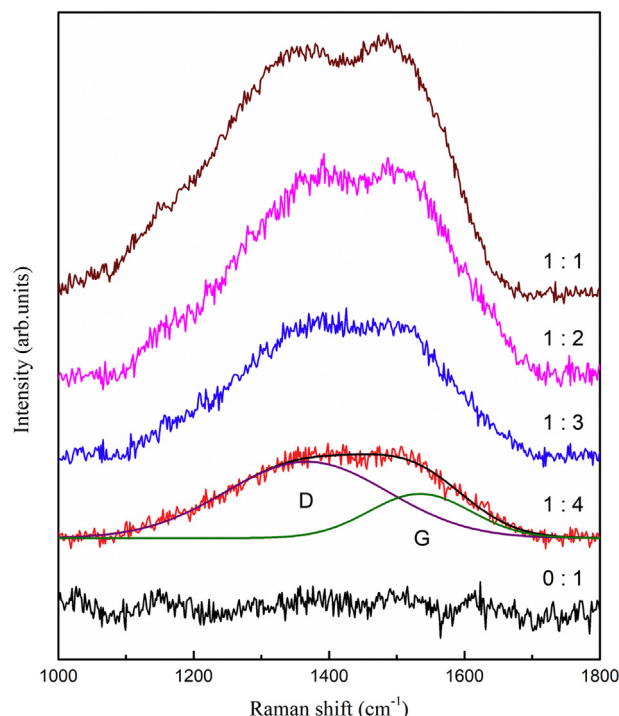


Fig. 5. Raman spectra of Ti-Al-Si-C-N coatings with different C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratios.

silicon carbide (a-SiC) or are incorporated in the fcc lattice, or both [41–46]. With an increase of C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio, the position of one of the major peaks evidently shifts from TiN (42.6°) to TiC (41.9°), indicating gradual replacement of TiN phase by TiC phase. This is consistent with the results in Table 2 showing that a decrease of N content with the increase of C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio. Those results clearly show that more and more N atoms in the Ti-N structure are being replaced by C atoms with the increase of C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio [33, 47].

To further understand the detailed structure of Ti-Al-Si-C-N coatings, the Raman spectra of the coatings were measured and the results are shown in Fig. 5. For the film deposited at C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio of 0:1, no Raman signal corresponding to amorphous carbon-based material was discovered. When C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio increased, a broad asymmetric band between 1000 and 1800 cm<sup>-1</sup> was present, which indicates the formation of amorphous carbon structure. Additionally, it can be clearly seen that as C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio increased the intensity of the amorphous carbon signal also increased. In order to perform a more detailed analysis, the broad asymmetric spectra were fitted by two Gaussian distributions. The band centered at about 1350 cm<sup>-1</sup> corresponding to the D band of the disordered structure and the other band centered at about 1580 cm<sup>-1</sup> related to the G band for the graphite structure [48–50]. The integrated intensity ratio of the D and G band (I<sub>D</sub>/I<sub>G</sub>) can be correlated with the sp<sup>3</sup>/sp<sup>2</sup> ratio. The I<sub>D</sub>/I<sub>G</sub> ratio of Ti-Al-Si-C-N coatings as a function of C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio is shown in Fig. 6. When C<sub>2</sub>H<sub>2</sub>/N<sub>2</sub> partial pressure ratio exceeded 1:4, the I<sub>D</sub>/I<sub>G</sub> value stayed around 4 ± 0.2, indicating the existence of a large portion of amorphous carbon phase consisted of sp<sup>2</sup> (graphite-like) bonds in the Ti-Al-Si-C-N coatings [51]. In summary, these results suggest that the Ti-Al-Si-C-N coatings have a nanocomposite structure composing of nanocrystallites (Ti,Al)(C,N) and amorphous phase. It should be noted, however, that the diffraction peaks of AlN crystalline phases with hexagonal closed packed structure were not observed from the XRD results, which indicated that the aluminum atoms are incorporated in the fcc lattice or exist as amorphous aluminum nitride (a-AlN), or both [52–54]. Further confirmation

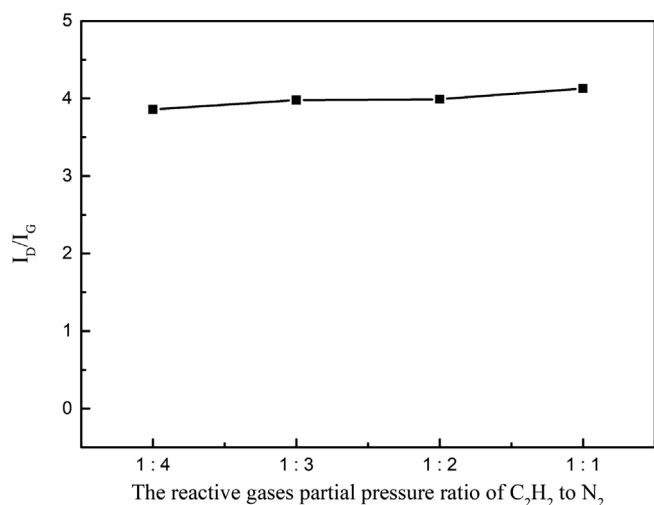


Fig. 6.  $I_D/I_G$  ratios of Ti-Al-Si-C-N coatings as a function of the  $C_2H_2/N_2$  partial pressure ratio.

of this interpretation would require an imaging with Transmission Electron Microscopy (TEM), which will be conducted in our follow-up study.

### 3.2. Mechanical properties of the Ti-Al-Si-C-N coatings

Fig. 7 shows the effect of  $C_2H_2/N_2$  partial pressure ratio on the hardness and Young's modulus of Ti-Al-Si-C-N coatings. As  $C_2H_2/N_2$  partial pressure ratio increased, the hardness evidently increased and reached a maximum value of 35 GPa at  $C_2H_2/N_2$  partial pressure ratio of 1:2. This can be explained that with the increase of  $C_2H_2/N_2$  partial pressure ratio, more and more carbon atoms replace nitrogen atoms to form Ti(C,N) solid solution. Because of the influence of the solid solution effect, the hardness of the coatings is expected to increase. Other possible reasons would be due to the nanocomposite structure of the nanocrystalline separated by amorphous phase mechanism would be the nanocomposite structure where the nanocrystalline (Ti,Al)(C,N) separated by amorphous phase (a-Si<sub>3</sub>N<sub>4</sub>, a-SiC and carbon). The nanocomposite structure can refine grain and hinder the movement of dislocations by the ideal interaction between nanocrystallites and the amorphous phase [36]. The highest hardness of Ti-Al-Si-C-N coatings demonstrates an optimal ratio of amorphous/nanocrystals and an appropriate  $C_2H_2/N_2$  partial pressure ratio (1:2 in this work). With the further increasing  $C_2H_2/N_2$  partial pressure ratio, the ideal interaction between nanocrystallites and the amorphous phase no longer exists, and therefore the hardness decreased [29]. The Young's modulus of Ti-

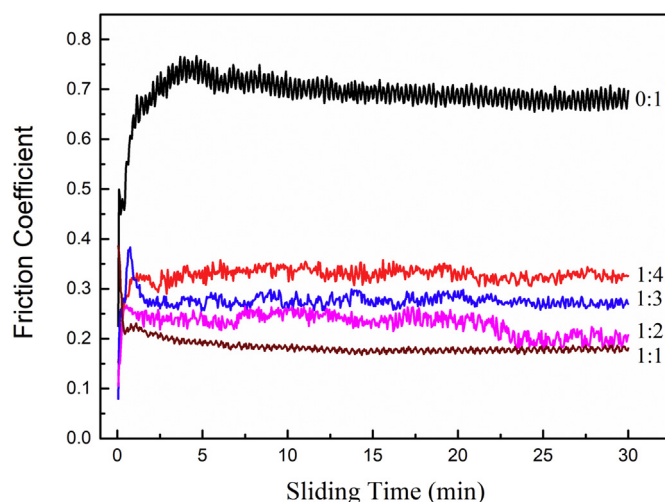


Fig. 8. Friction coefficients curves of the Ti-Al-Si-C-N coatings with different  $C_2H_2/N_2$  partial pressure ratios.

Al-Si-C-N coatings is also shown in the same plot in Fig. 7. As  $C_2H_2/N_2$  partial pressure ratio increased from 0:1 up to 1:4, Young's modulus, which is sensitive to density and atomic structure of coating [55], sharply decreased from 360 GPa to 290 GPa. This decrease could be explained by the different chemical compositions in those coatings. When the carbon atoms accompanied with hydrogen were introduced in to the coatings, flexible  $sp^1$ ,  $sp^2$ ,  $sp^3$  sites were formed, coinciding with the D and G peak seen in Fig. 5. As  $C_2H_2/N_2$  partial pressure ratio increased from 1:4 up to 1:2, Young's modulus dramatically increased and reached the maximum values of 380 GPa. This result would be attributed to the densification of Ti-Al-Si-C-N coatings with a-Si<sub>3</sub>N<sub>4</sub> and a-C filling the open structure of Ti(C,N) grain boundary. When  $C_2H_2/N_2$  partial pressure ratio changed from 1:2 to 1:1, the Young's modulus renewedly decreased again. It may be due to the increase of volume fraction of a-Si<sub>3</sub>N<sub>4</sub> and a-C, which has a relatively lower atomic density than that of the crystalline Ti(C,N) phase [29].

The friction coefficients curves of the Ti-Al-Si-C-N coatings with different  $C_2H_2/N_2$  partial pressure ratios are shown in the in Fig. 8. It can be seen that the Ti-Al-Si-C-N coatings generally have a lower friction coefficient, comparing with that of TiAlSiN coatings. With the increase of  $C_2H_2/N_2$  partial pressure ratio, the friction coefficients of the Ti-Al-Si-C-N coatings decreased remarkably from about 0.3 to approximately 0.17. The reduction of friction coefficient is due to the formation of the new amorphous phase (a-C) (supported by the results from Raman spectroscopy) and the increase of carbon content in the coatings. Amorphous carbon is known to exhibit low friction coefficient and can be used to as a solid lubricant [56]. A higher value of  $I_D/I_G$  ratio corresponds to higher  $sp^2$  content in materials with good solid self-lubricant properties. Therefore, with the increase of  $C_2H_2/N_2$  partial pressure ratio, the amount of a-C increases, which in turn, decreases the friction coefficients of the coatings.

To investigate the wear resistance of the Ti-Al-Si-C-N coatings against GC15 balls, the surface morphologies of the wear tracks were observed by laser confocal scanning microscope. The 3D images and 2D cross-sectional profiles of the wear tracks on the Ti-Al-Si-C-N coatings with different  $C_2H_2/N_2$  partial pressure ratios are shown in Fig. 9, in which the maximum wear depth was marked. It is important to point out that the scale ranges of the Y-axis in Fig. 9(a) are much larger than that in the other figures due to the very width of the wear track. The wear track on TiAlSiN coatings is much wider (about 430  $\mu$ m), and deeper (2.7  $\mu$ m) when compared with that on Ti-Al-Si-C-N coatings. The maximum depths of the wear track for the coatings with the  $C_2H_2/N_2$  partial pressure ratio of 0:1, 1:4, 1:3, 1:2, and 1:1 are 2.7, 2.5, 2.1, 1.2, and 0.7  $\mu$ m, respectively. Fig. 10 shows the wear rates of the Ti-Al-Si-C-

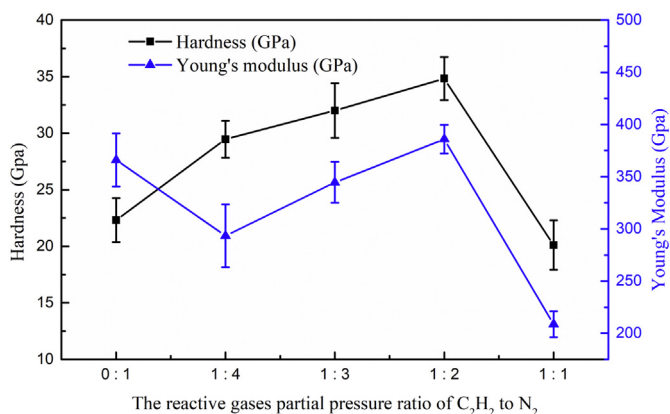
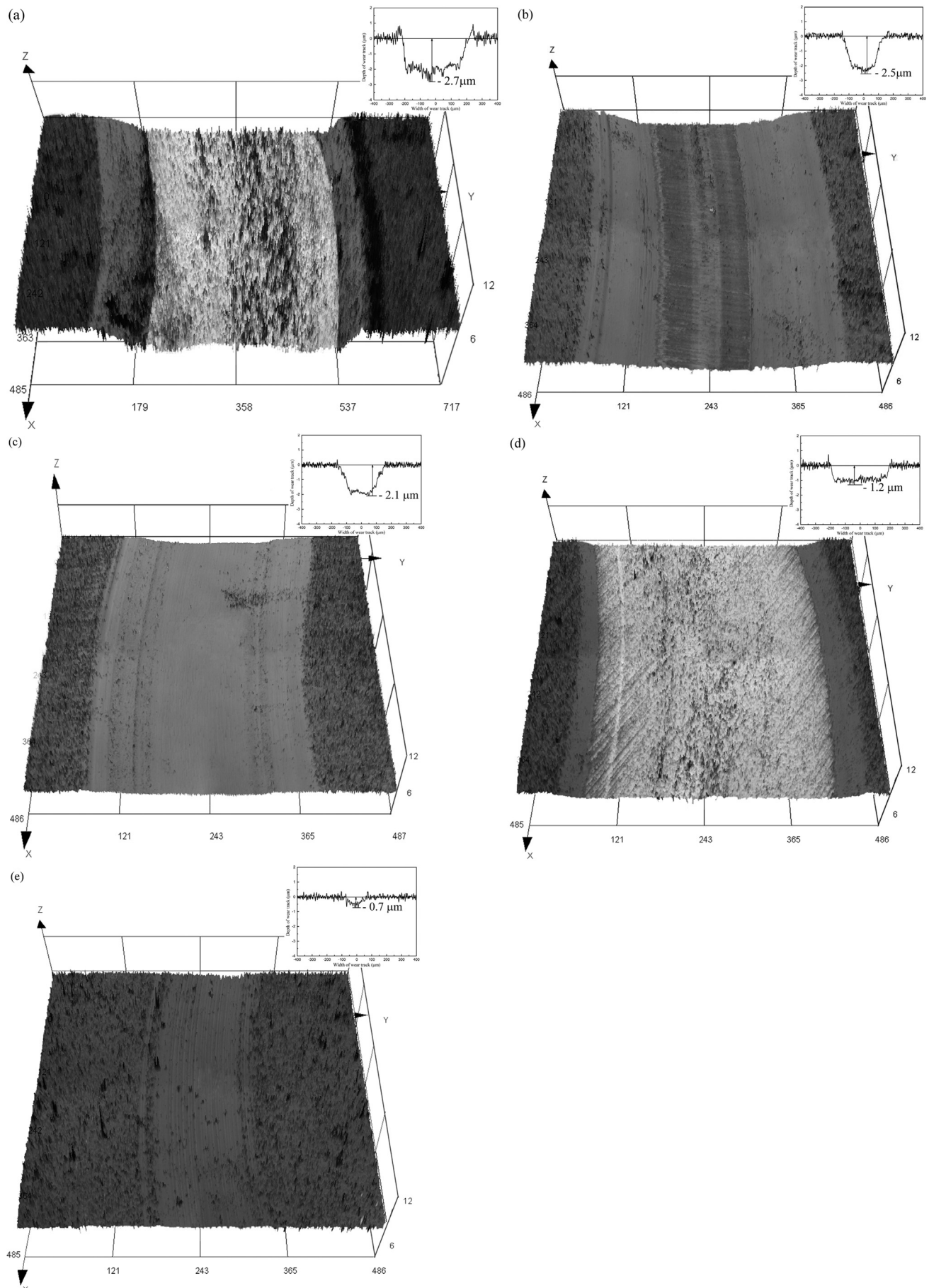
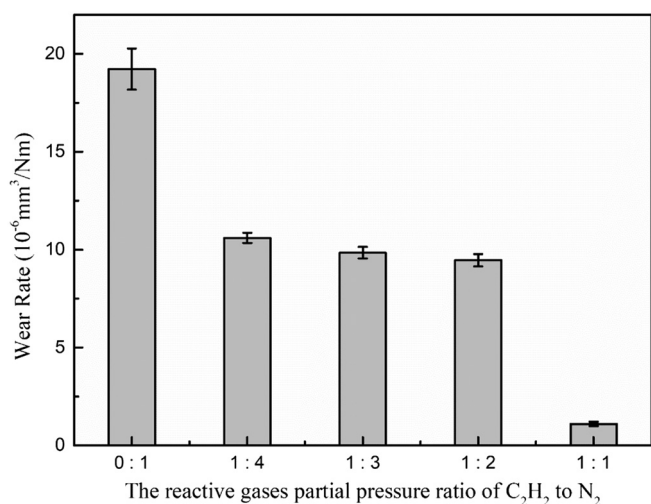


Fig. 7. Effect of  $C_2H_2/N_2$  partial pressure ratio on the hardness and Young's modulus of Ti-Al-Si-C-N coatings.



(caption on next page)

**Fig. 9.** The 3D images and 2D cross-sectional profiles of the wear tracks on the Ti-Al-Si-C-N coatings with different  $C_2H_2/N_2$  partial pressure ratios: (a) 0:1 (b) 1:4 (c) 1:3 (d) 1:2 (e) 1:1.



**Fig. 10.** Wear rates of the Ti-Al-Si-C-N coatings with different  $C_2H_2/N_2$  partial pressure ratios.

N coatings with different  $C_2H_2/N_2$  partial pressure ratios. The wear rates were calculated from the cross-sectional profiles of wear tracks. For TiAlSiN coatings, the wear rate is about  $1.92 \times 10^{-5} \text{ mm}^3/\text{Nm}$ . As the  $C_2H_2/N_2$  partial pressure ratio increased from 0:1 to 1:4, the wear rate reduced by 43% to  $1.02 \times 10^{-5} \text{ mm}^3/\text{Nm}$ . A further increase of the  $C_2H_2/N_2$  partial pressure ratio to 1:1 leads to a gradual decrease in the wear rate to  $1.1 \times 10^{-6} \text{ mm}^3/\text{Nm}$ . Those results indicate a continuous improvement of wear resistance of the coatings with an increase of  $C_2H_2/N_2$  partial pressure ratio.

The wear resistance is strongly dependent on the microstructure, hardness of the coating, friction coefficient, and lubricated phase [56]. For TiAlSiN coatings, the combination of the relatively looser microstructure, low hardness of the coating, high friction coefficient, and absence of the lubricated phase in the coating contribute to the high wear rate. When  $C_2H_2/N_2$  partial pressure ratio increased from 0:1 to 1:2, the combination of densified microstructure, increase of the hardness and formation of a lubricated phase (a-C) in the Ti-Al-Si-C-N coatings resulted a remarkable decrease in the wear rate. With a further increase of  $C_2H_2/N_2$  partial pressure ratio, the reduction in the friction coefficient causes a gradual decrease in the wear rate.

#### 4. Conclusion

In this study, Ti-Al-Si-C-N coatings were prepared by reactive magnetron sputtering technique with different  $C_2H_2/N_2$  partial pressure ratios. The effect of  $C_2H_2/N_2$  partial pressure ratio on coatings' microstructure and mechanical properties has been investigated. From XRD and Raman studies it could be suggested that the coatings were composed of a mixture of nanocrystallinities (Ti,Al)(C,N) and amorphous phase. With the increase of  $C_2H_2/N_2$  partial pressure ratio, the columnar crystals structure in the coatings was restrained, resulting a densified microstructure. The deposition rate dramatically increased from 88.33 nm/min to about 133 nm/min as the  $C_2H_2/N_2$  partial pressure ratio increased from 0:1 to 1:2, then decreased with the further increase of  $C_2H_2/N_2$  partial pressure ratio. As  $C_2H_2/N_2$  partial pressure ratio increased, the hardness remarkably increased and reached a maximum value of 35 GPa at  $C_2H_2/N_2$  partial pressure ratio of 1:2 before decreasing. Moreover, both the friction coefficients and wear rates sharply decreased as  $C_2H_2/N_2$  partial pressure ratio increased. The lowest friction coefficient and wear rate were 0.17 and

$1.1 \times 10^{-6} \text{ mm}^3/\text{Nm}$ . This behavior would be attributed to the densified microstructure, the significant increase of the hardness and the formation of a lubricated phase (a-C) in the Ti-Al-Si-C-N coatings. Our comprehensive study on the effect of  $C_2H_2/N_2$  provides insights in further improving the properties of Ti-Al-Si-C-N coatings.

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